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Determination of Kjeldahl-Nitrogen



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Application

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Use

The method is suitable for samples which contents nitrate or protein. The sample is digested with sulfuric acid in the presence of a catalyst. This converts organically bound nitrogen to ammonium sulfate.

In a distillation apparatus the digestion solution is treated with NaOH and the released ammonia (NH $_3$) is distilled into a solution of boric acid. This solution is then titrated with an acid titrant (HCl or H $_2$ SO $_4$) to a pH endpoint of 4.65.

Appliances

- Titrator: TL 6000/7000 (TL 6000/7000 M2/20) consists of
- Basic device
- Magnetic stirrer TM 235
- 20 mL Exchange unit WA 20, with amber glass bottle for the titrant, complete
- pH combination electrode A 162 DIN ID

Electrodes

Electrode: A 162 DIN ID

Calibration: DIN buffer pH= 4.00 and pH= 7.00



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Reagents

- Titrant: HCl 0.01 mol/ 1 mol/l or H₂SO₄ 0.005 0.5 mol/l
- Titer determination: TRIS or potassium carbonate

Description

Calibration

The pH combination electrode is calibrated in technical buffer pH=4.00 and pH= 7.00 or in DIN buffer pH= 4.01 and pH= 6.87.

Example of the calibration documentation:

Calibration

Buffers used

pH buffer 1: TEC_4.000 pH buffer 2: TEC_7.000

Measured values

pH buffer 1: TEC_4.000 165.6 mV / 23.4 °C pH buffer 2: TEC_7.000 -11.2 mV / 23.0 °C

Calibration data

 Slope:
 99.4 % / -58.8 mV/pH

 Zero point:
 pH 6.81 / -11.2 mV

 Temperature:
 23.4 °C (a)

 Date and time:
 07.03.13 / 15:04



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Determination of the exact concentration of the standard solution (0.1 mol/l /0.05 mol/l)

The exact concentration of the acid titrant can be determined using a standard TRIS (recommended) or potassium carbonate. The TRIS is dried for 24 hours in a desiccator at room temperature.

In a 150 mL beaker, 0.15 g TRIS are weighed accurately and dissolved in 80 mL of dist. water with stirring. It is titrated with 0.1 mol/l sodium hydroxide solution.



Pic. left: titer



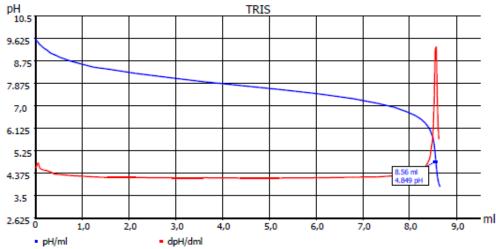
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GLP documentation

Titration graph



Method data

Method name: Titre HCl Titration duration: 3 m 8 s
End date: 13.09.12 End time: 14:39:30

Titration data

 Sample ID:
 TRIS
 Weight:
 0.1038 g

 Start pH:
 pH 9.590
 End pH:
 pH 3.864

 Start temperature:
 25.0 °C (m)
 End temperature:
 25.0 °C (m)

Zero point: pH 6.83 / -10.0 mV Slope: 100.6 % / -59.5 mV/pH

EQ: 8.560 ml / pH 4.849 Titre: 0.1001 mol/l

Calculation formula

Titre: (W*F2)/((EQ1-B)*M*F1) -> M103

Mol (M): 121.14000

Weight (W): man Factor 2 (F2): 1000.0000 Blank value (B): 0.0000 ml Factor 1 (F1): 1.0000

Statistics: Off



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Page 2: Method parameters Titer determination:

Method data overall view

Method name: Titre HCl Created at: 09/13/12 14:23:02
Method type: Automatic titration Last modification: 09/13/12 14:27:56

Measured value: pH Damping settings: None Titration mode: Dynamic Documentation: GLP

Dynamic: Steep

Measuring speed / drift: Normal: minimum holding time: 02 s

maximum holding time: 15 s

Measuring time: 02 s

Drift: 20 mV/min

Initial waiting time: 0 s
Titration direction: Decrease
Pretitration: Off
End value: 2.500 pH
EQ: On (1)

Slope value: Steep Value: 700

Dosing parameter

Dosing speed: 100 % Filling speed: 30 s

Maximum dosing volume: 50.00 ml

Unit values

 Unit size:
 20ml

 Unit ID:
 10039005

 Reagent:
 HCl 0.1 mol/L

 Batch ID:
 no Charge

 Concentration [mol/l]:
 0.10070

Determined at: 12/05/11 19:18:45

Expire date: 08/18/12
Opened/compounded: 09/10/11
Test according ISO 8655: 05/10/11

Last modification: 09/13/12 14:35:18

Device information

Device: TitroLine 7000 Serial number: 00012

Software version: 1230 Titre_HCl_13_09_12-14_36_21.pdf



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Application

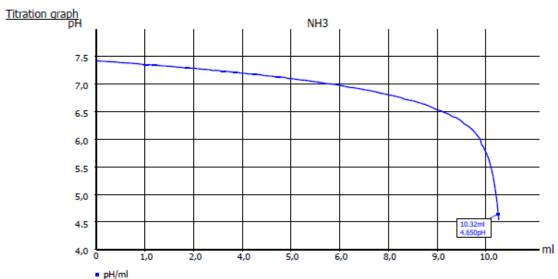
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Titration of the sample

When the distillation is finish the beaker (250 ml or bigger) is placed on the magnetic stirrer. The electrode and burette tip are immersed into the solution, the Kjeldahl method is selected and started. The titrator titrate until an endpoint of pH 4.65 is reached. A blank titration with more carefully titration parameters should be also carried out before to find a blank value for the Kjeldahl instrument.

Result example:

Standard documentation



Method data

Method name: Kjehldal End date: 30.11.11

End time: 16:37:56 Titration duration: 2 m 31 s

Titration data

Sample ID: NH3 Pattern: 1.5500ml

 Start pH/temp:
 pH7.420
 End pH/temp:
 pH4.529

 start temperature:
 25.0 °C (m)
 end temperature:
 25.0 °C (m)

Zero point: pH 6.82 / 0.1 mV Slope: 98.4% / -58.2 mV/pH

EP1: 10.317 ml/pH4.650 %s

Nitrogen: 9.325 g/l

Calculation formula

Formulas (EP1-B)*T*M*F1/(V*F2) Factor (F1): 2.0000

Factor (F2): 1.0000 Factor (F3): 1.0000

Blank value (B): 0.0000ml

Titre (T): 0.05000000(f)

Mol (M): 14.01000



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Linear steps:

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Method

Method data

Kjeldahl Method name: Created at:

0.040 ml

04/23/13 15:24:04 Method type: Automatic titration Last modification: 05/16/13 12:54:59 Measured value: pН

Titration mode: End pt. Documentation: GLP

Measuring speed / drift: minimum holding time: 02 s Normal:

maximum holding time: 15 s Measuring time: Drift: 20 mV/min

Initial waiting time: 0 s Titration direction: Decrease Pretitration: Off

Endpoint 1: pH 4.650 delta endpoint 1: pH 1.200

Endpoint delay 1: 3 s Endpoint 2: Off

Dosing parameter

Dosing speed: 60.00 % Filling speed: 30 s

Maximum dosing volume: 50.00 ml

Calculation formula

Nitrogen: (EP1-B)*T*M*F1/(W*F2) Mol (M): 14.01000

Unit: Decimal places:

Blank value (B): Titre (T): auto Factor 1 (F1): 0.1000 Weight (W): man Factor 2 (F2): 1.0000 Statistics: Off

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Blank method:

Method data

 Method name:
 Kjeldahl blank
 Created at:
 05/16/13 12:56:24

 Method type:
 Automatic titration
 Last modification:
 05/16/13 12:57:33

Measured value: pH

Titration mode: End pt. Documentation: GLP

Linear steps: 0.010 ml

Measuring speed / drift: Normal: minimum holding time: 02 s

maximum holding time: 15 s

Measuring time: 02 s

Drift: 20 mV/min

Initial waiting time: 0 s
Titration direction: Decrease
Pretitration: Off

Endpoint 1: pH 4.650 delta endpoint 1: pH 1.200

Endpoint delay 1: 3 s

Endpoint 2: Off

Dosing parameter

Dosing speed: 30.00 % Filling speed: 30 s

Maximum dosing volume: 50.00 ml

Calculation formula

blank value: EP1 -> M02

Unit: ml Decimal places: 3

Statistics: Off

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Notes

If you have any questions on the application, you can feel free to contact us..



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